A STEREOCONTROLLED TOTAL SYNTHESIS OF (±)-RENIERAMYCIN A

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Abstract: The first total synthesis of (\pm) -renieramycin A (1a) is described. The stereochemistry of the angelate side chain was unequivocally determined by X-ray crystallographic analysis of the penultimate intermediate.

The renieramycins (1) were isolated in minute quantities from the bright blue marine sponge *Reniera* sp. by Faulkner and co-workers.¹ These antimicrobial metabolites belong to a growing family of isoquinolinequinone antibiotics, that includes the more notable saframycin A (2a) which shows potent antitumor activities.² Although limited supply of the renieramycins precludes extensive biological testing, structural similarities to the saframycins suggest the potential for antitumor activity. The stereochemistry of the angelate side chain of renieramycins was originally assigned as α based on difference NOE studies.^{1a} However, striking resemblance of the high-field ¹H NMR spectrum of saframycin C (2c) to that of renieramycin A (1a) strongly suggests that both antibiotics share the same β -configuration at the side chains,³ as revised recently by Faulkner.^{1b} As a part of our ongoing synthetic endeavor in the area of saframycins,⁴ we have been pursuing a total synthesis of renieramycin A (1a). Herein we report the first total synthesis of 1a, that unambiguously proved the stereochemistry of the side chain.



The present synthesis followed closely the pathway established for saframycin A $(2a)^{4a,b}$ except that the specially protected benzaldehyde 7 was employed to set up the critical benzylic oxidation. As shown in Scheme 1, 7 was prepared from the readily available aldehyde 3^5 in 55% overall yield. Selective demethylation of **3** with BCl₃ followed by allylation gave **4**. Baeyer-Villiger oxidation of aldehyde **4** was effected by 30% H₂O₂ and a catalytic amount of SeO₂ to avoid epoxidation of the olefin. Methanolysis of the resultant formate and protection of the phenol as MOM ether furnished **5**. Claisen rearrangement of **5** at 210 °C, followed by base-catalyzed isomerization of the olefin, gave the conjugated olefin **6**. The phenol **6** was then protected as sturdy 3-hydroxypropyl ether. Further protection of the primary alcohol as dimethythexylsilyl (DMTS) ether and subsequent ozonolysis of the olefin yielded the aldehyde **7**. Condensation of **7** with N,N'-diacetylpiperazinedione was achieved by treatment with *t*-BuOK/*t*-BuOH in THF at 0 °C, giving exclusively the Z-isomer **8** (86%).⁶ After catalytic



(a) BCI₃, CH₂CI₂, 0 °C. (b) allyl bromide, K₂CO₃, DMF, 80 °C. (c) 30% H₂O₂, cat. SeO₂, *t*BuOH, 40 °C. (d) Et₃N, MeOH, 23 °C. (e) MeOCH₂CI, *i*-Pr₂NEt, CH₂CI₂, 80 °C. (f) PhNEt₂, 210 °C. (g) KOH, DMSO, 105 °C. (h) Br(CH₂)₃OH, Nal, K₂CO₃, DMF, 70 °C. (i) DMTSCI, *i*-Pr₂NEt, DMF, 70 °C. (j) O₃, MeOH, -78 °C; Me₂S. (k) N,N'-diacetylpiperazinedione, *t*-BuOK, THF, 0 °C. (l) H₂, Pd/C, EtOAc. (m) CbzCl, Et₃N, DMAP, CH₂CI₂, -30°C. (n) 5-benzyloxy-2,4-dimethoxy-3-methylbenzaldehyde, *t*-BuOK, THF, -78 °C; DBU, 0 °C. (o) NaBH₄, AcOH, EtOH, 0 °C. (p) HCO₂H, 55 °C. (q) NaOH, MeOH, 23 °C.

hydrogenation of olefin 8, the piperazinedione ring was activated again by introduction of a carbobenzoxy group. Condensation of 9 with 5-benzyloxy-2,4-dimethoxy-3-methylbenzaldehyde^{4b} furnished 10 in 68% yield from 8 (t-BuOK, THF, -78 °C, then DBU, 0 °C). Selective reduction of the ring carbonyl with NaBH₄, acyliminium ion-mediated cyclization in formic acid, and hydrolysis of the resulting formate gave the desired cyclized product 11 in 68% yield.

Catalytic hydrogenation of olefin 11 over Raney nickel took place from the less hindered exo side to give the desired endo-compound 12 as the sole product (70%) (Scheme 2). Since benzylic oxidation of 14 with DDQ gave an intractable mixture when R was methyl, we



(a) H₂ (1200 psi), Ra-Ni (W2), EtOH, 80 °C. (b) CICO₂(CH₂)₂CO₂Me, PhNMe₂, CH₂Cl₂, 23 °C. (c) BnBr, K₂CO₃, DMF, 23 °C. (d) Swern oxid. (e) DDQ, THF/H₂O (8:1), 23 °C. (f) Mel, K₂CO₃, DMF, 23 °C. (g) DBU, MeOH, 23 °C. (h) HCHO, NaBH₃CN, TFA, MeOH, 23 °C. (i) AlH₃, THF, 23 °C. (j) H₂, Pd/C, EtOH, 23 °C. (k) glycolaldehyde angelate, CH₃CN, 50 °C. (l) DDQ, acetone/H₂O (20:1), 23 °C.

protected the bridgehead amine of 12 as the base-sensitive urethane $(ClCO_2(CH_2)_2CO_2Me,^7$ PhNMe₂). Subsequent benzylation of the phenols gave 13 (58% from 12). Swern oxidation⁸ of the primary alcohol 13 provided directly the desired phenol 14 in 78% yield, setting the stage for the critical oxidation. Upon treatment with DDQ in aqueous THF, 14 gave exclusively the desired alcohol.⁹ Methylation of the phenol, deprotection of the urethane with DBU through retro-Michael reaction, and reductive methylation of the resultant amine afforded 15 in 45% overall yield from 14. Reduction of lactam 15 with AlH₃ followed by hydrogenolysis of the benzyl ethers gave amino phenol 16 (64%). Pictet-Spengler reaction of 16 with glycolaldehyde angelate¹⁰ in CH₃CN at 50 °C furnished a 5:1 mixture of 17 and its minor α -epimer in 66% yield. The structure of 17 was determined by a single crystal X-ray analysis. Finally, 17 was oxidized with DDQ to give (±)-renieramycin A (1a) in 48% yield. The ¹H NMR spectrum of the synthetic renieramycin A was identical with that of the authentic sample.¹¹

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References and Notes

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